

DOUBLE DISPLACEMENT-2



MATERIAL REQUIRED



IT IS COLOURLESS ON DISSOLVING



ADD SOME SODIUM SULPHATE IN BEAKER

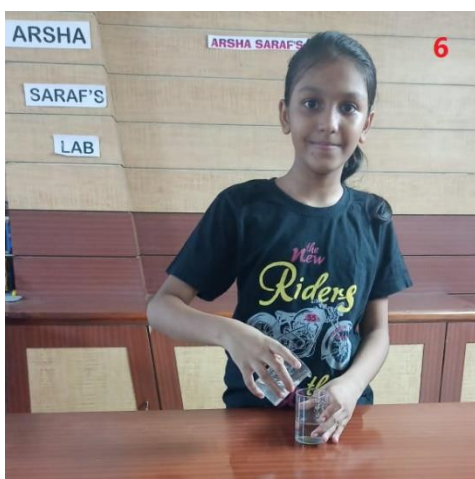


ADD BARRIUM CHLORIDE IN ANOTHER BEAKER

ARSHA SARAF'S LAB



IT TURNS MILKY ON DISSOLVING



MIX TWO SOLUTIONS IN ANOTHER BEAKER



AFTER SOMETIME, WHITE PRECIPITATE OF BARIUM SULPHATE IS DEPOSITED AT THE BOTTOM AND SODIUM CHLORIDE IS DISSOLVED IN WATER

ARSHA SARAF'S LAB

$\text{BaCl}_2 + \text{Na}_2\text{SO}_4 \rightarrow \text{BaSO}_4 + \text{NaCl}$
This is a double displacement reaction